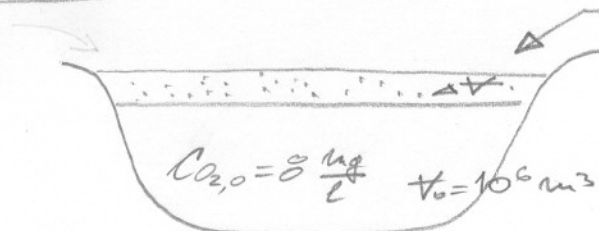


### Zadatak 3 (Zatvoreno)

$$Q_{ul} = 120 \text{ l/s}$$

$$\Delta t = 2 \text{ dana}$$



početna masa rastv. kiseonika u jezercu:

$$M_{O_2,0} = 8 \frac{\text{mg}}{\text{l}} \times 10^6 \text{ m}^3 \times 10^3 \frac{\text{l}}{\text{m}^3} = 8000 \text{ kg}$$

- ulaz mase organske materije ( $C_6H_{12}O_6$ ):

$$M_{OM} = \underbrace{120 \frac{\text{l}}{\text{s}} \times 200 \frac{\text{mg}}{\text{l}}}_{Q_{ul,OM}} \times \underbrace{2 \times 86400 \text{ s}}_{\Delta t} = 4147.2 \text{ kg } C_6H_{12}O_6$$

u ekvivalentima rastvorjenog  $O_2$ :  $M_{OM} = 1.07 \times M_{OM} = 4437.5 \text{ kg } O_2$

- maseni ulaz amonijum jona ( $NH_4^+$ )

$$M_{NH_4^+} = 120 \frac{\text{l}}{\text{s}} \times 30 \frac{\text{mg}}{\text{l}} \times 2 \times 86400 \text{ s} = 622.08 \text{ kg } NH_4^+$$

u ekvivalentima  $O_2$ :  $M_{NH_4^+} = 3.56 \frac{\text{kg } O_2}{\text{kg } NH_4^+} \cdot 622.08 = 2214.6 \text{ kg } O_2$

- masa  $O_2$  nakon razgradnje ( $t = t_1$ )

$$M_{O_2,1} = M_{O_2,0} - M_{OM} - M_{NH_4^+}$$

$$M_{O_2,1} = 8000 \text{ kg} - 4437.5 \text{ kg} - 2214.6 \text{ kg} = 1348 \text{ kg}$$

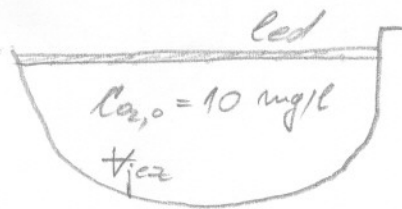
- ukupna zapremina jezera:

$$V_1 = V_0 + Q_{ul} \cdot \Delta t = 1020736 \text{ m}^3$$

- koncentracija  $O_2$  nakon razgradnje:

$$C_{O_2,1} = \frac{M_{O_2,1}}{V_1} = 1.32 \frac{\text{mg}}{\text{l}}$$

### Zadatak 4



$$r = 1.4 \frac{\text{mg}}{\text{l} \cdot \text{ned}}$$

hipoksični uslovi:  $C_{O_2} < 2 \text{ mg/l}$   
 - potrošnja  $O_2$  (respiratorna)

$$Q_{m,r} = r \cdot V_{ice}$$

- potrošnja  
 j-na održanja mase  $O_2$ :  $M_1 = M_0 - M_r$

$$\underbrace{C_{O_2,1}}_{2 \text{ mg/l}} \cdot V_{ice} = C_{O_2,0} \cdot V_{ice} - r \cdot V_{ice} \cdot t$$

$$2 \text{ mg/l} = 10 \text{ mg/l} - 1.4 \frac{\text{mg}}{\text{l} \cdot \text{ned}} \cdot t$$

$$t = 5.7 \text{ nedelja} = 5 \text{ nedelja i } 5 \text{ dana}$$

Za 5 nedelja i 5 dana nastaje hipoksični uslovi

### Zadatak 5 (otvoreni reaktor sa potpunim mešanjem)

#### Zadatak 5

$$Q_{ul} = 20 \text{ l/s}$$

$$Q_{izl} = 20 \text{ l/s}$$

$$C_{O_2} = C_{izl}$$

potpuno mešanje

$$C_{ul, O_2} = 6 \frac{\text{mg}}{\text{l}}$$

$$C_{izl, O_2} = 10 \text{ mg/l}$$

$$V = 2000 \text{ m}^3$$

$r$  - respiracija + raspadanje org. mat.  
 $r = 0.2 \text{ mg/(l} \cdot \text{s)}$

$$\text{ulaz} - \text{izlaz} + \text{produkcija} - \text{potrošnja} = 0$$

$$Q_{m,ulaz} - Q_{m,izlaz} + Q_{m,aer} - Q_{m,r} = 0 ; \quad Q_{m,r} = r \cdot V$$

$$20 \frac{\text{l}}{\text{s}} \times 6 \frac{\text{mg}}{\text{l}} - 20 \frac{\text{l}}{\text{s}} \times 10 \frac{\text{mg}}{\text{l}} + Q_{m,aer} - 0.2 \frac{\text{mg}}{\text{l}} \cdot \frac{1}{3600 \text{ s}} \times 2 \times 10^6 \text{ l} = 0$$

$$Q_{m,aer} = 247 \frac{\text{mg}}{\text{s}} = 247 \frac{\text{mg}}{\text{s}} \times \frac{10^6 \frac{\text{kg}}{\text{t}}}{3600 \frac{\text{h}}{\text{s}}} = 0.9 \text{ kg } O_2/\text{h}$$

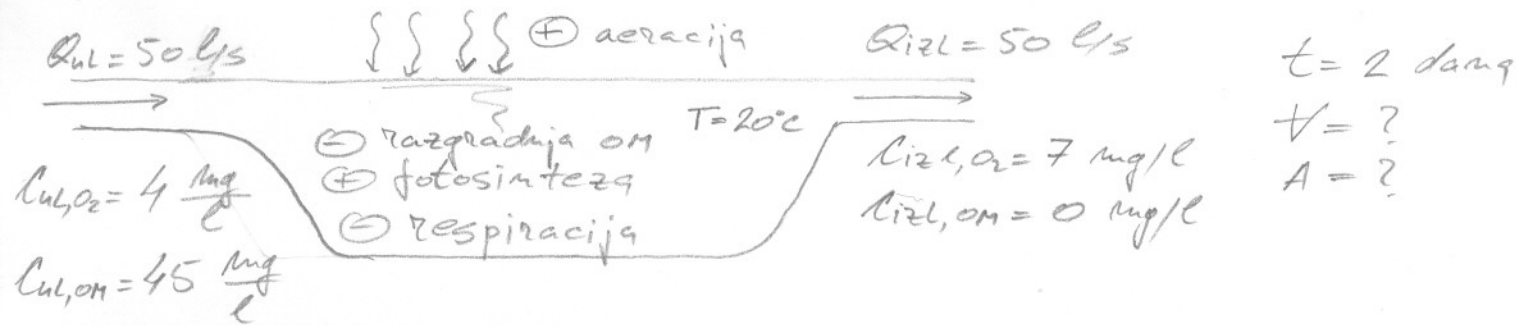
Aeraciju se postiže ovazdušavanjem

$$Q_{m, vazduha} = \frac{0.9 \text{ kg } O_2/\text{h}}{20\%} = 4.44 \text{ kg vazduha/h}$$

$$Q_{vazduha} = \frac{Q_{m, vazduha}}{\rho_{vazduha}} = 4.44 \text{ m}^3 \text{ vazduha/h}$$

(2)

Zadatak 6 (otvoreni reaktor sa potpunim mešanjem)



$K_a = 0.34 \times 10^{-3} \frac{\text{m}}{\text{s}}$ ;  $r = 2.4 \frac{\text{mg}}{\text{l.dan}}$ ;  $f = 6.2 \frac{\text{mg}}{\text{l.dan}}$

- Jednačinu održavanja mase  $O_2$ :

$$\underbrace{\text{ulaz}}_{(1)} - \underbrace{\text{izlaz}}_{(2)} + \underbrace{\text{produkcija}}_{(3)} - \underbrace{\text{potrošnja}}_{(4)} = 0 \quad \left[ \frac{M}{T} \right]$$

potrebna zapremina:  $V = Q_{ul} \cdot 2 \text{ dana} = 8640 \text{ m}^3$

(1) ulaz:  $Q_{m,ul} = 4 \frac{\text{mg}}{\text{l}} \cdot 50 \frac{\text{l}}{\text{s}} = 200 \frac{\text{mg}}{\text{s}}$

(2) izlaz:  $Q_{m,izl} = 7 \frac{\text{mg}}{\text{l}} \times 50 \frac{\text{l}}{\text{s}} = 350 \frac{\text{mg}}{\text{s}}$

(3) produkcija

(3.1) fotosinteza:  $Q_{m,f} = f \cdot V = 6.2 \frac{\text{mg}}{\text{l.dan}} \cdot \frac{1 \text{ dan}}{86400 \text{ s}} \times 8640 \times 10^3 \text{ l}$

$Q_{m,f} = 620 \text{ mg/s}$

(3.2) aeracija:  $Q_{m,aer} = K_a \cdot A \cdot (C_{sat} - C_{O_2})$ ;  $A = ?$

$T = 20^\circ\text{C}$  pa je  $C_{O_2,sat} = 9.2 \text{ mg/l}$

$C_{O_2} = C_{izl,O_2} = 7 \text{ mg/l}$

(4) potrošnja

(4.1) razgradnja organske materije

u ekvivalentima  $O_2$ :  $Q_{m,OM} = r_{og} \cdot C_{ul,OM} \cdot Q_{ul} = 2408 \text{ mg/s}$

(4.2)  $Q_{m,OM} = 1.07 \frac{\text{mg } O_2}{\text{mg gl.}} \times 45 \frac{\text{mg}}{\text{l}} \times 50 \frac{\text{l}}{\text{s}} = 2408 \text{ mg/s}$

(4.2) respiracija  $Q_m = r \cdot V$

$Q_m = 2.4 \frac{\text{mg}}{\text{l.dan}} \times \frac{1 \text{ dan}}{86400 \text{ s}} \times 8640 \times 10^3 \text{ l} = 240 \text{ mg/s}$

## Zadatok 6 (nastavak)

$$Q_{m,ul} - Q_{m,izl} + Q_{m,t} + Q_{m,aer} - Q_{m,OH} - Q_{m,r} = 0$$

$$200 \frac{\text{mg}}{\text{s}} - 350 \frac{\text{mg}}{\text{s}} + 620 \frac{\text{mg}}{\text{s}} + Q_{m,aer} - 2408 \frac{\text{mg}}{\text{s}} - 240 \frac{\text{mg}}{\text{s}} = 0$$

$$Q_{m,aer} = 2178 \frac{\text{mg}}{\text{s}} ; \quad Q_{m,aer} = k_a \cdot A \cdot (C_{sat} - C_{or})$$

$$0.34 \times 10^{-3} \frac{\text{kg}}{\text{s}} \cdot A \cdot (9.2 - 7) \frac{\text{mg}}{\text{l}} = 2178 \frac{\text{mg}}{\text{s}}$$

$$0.34 \times 10^{-3} \frac{\text{kg}}{\text{s}} \times A \times 2.2 \times 10^3 \frac{\text{mg}}{\text{m}^3} = 2178 \frac{\text{mg}}{\text{s}}$$

$$A = 2912 \text{ m}^2$$

## Zadatok 7

$$BPK_5 = 250 \text{ mg/l}$$

$$t = 5 \text{ danq}$$

$$k = 0.15 \text{ danq}^{-1}$$

$$BPK_5 = L_0(1 - e^{-k \cdot 5})$$

$$L_0 = 474 \text{ mg/l}$$

## Zadatok 8

$$Q_{or} = 500 \frac{\text{l}}{\text{s}}$$

$$BPK_{5,or} = 350 \frac{\text{mg}}{\text{l}}$$



$$Q_r = 10 \frac{\text{m}^3}{\text{s}}$$

$$C_{O_2,r} = 8 \text{ mg/l}$$

$$BPK_{5,r} = 3 \text{ mg/l}$$

$$C_{O_2}(x=?) = 2 \text{ mg/l}$$

$$v = 0.5 \text{ m/s}$$

$$k = 0.35 \text{ danq}^{-1}$$

$$C_{O_2,mizv} = 7.6 \text{ mg/l}$$

$$BPK_{5,mizv} = 19.5 \text{ mg/l}$$

$$L_{0,mizv} = \frac{BPK_{5,mizv}}{(1 - e^{-0.35 \times 5})} = 23.6 \frac{\text{mg}}{\text{l}}$$

$$C_{O_2} = C_{O_2,mizv} - L_{0,mizv}(1 - e^{-0.35 \cdot t})$$

$$L_{0,mizv}(1 - e^{-0.35 \cdot t}) = (7.6 - 2) \frac{\text{mg}}{\text{l}}$$

$$t = 0.77 \text{ danq}$$

$$x = 33.2 \text{ km}$$